

The Physics of Energy

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I Introduction to Energy

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Energy

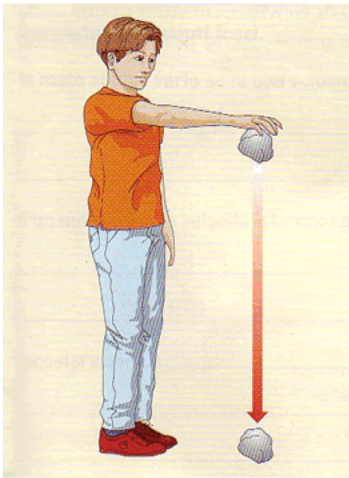


What is it?

Energy

In physics, energy (Ancient Greek: ἐνέργεια *energeia* "activity, operation") is an indirectly observed quantity that is often understood as the **ability of a physical system to do work on other physical systems**

Capability of doing WORK... $WORK = FORCE \times Displacement$



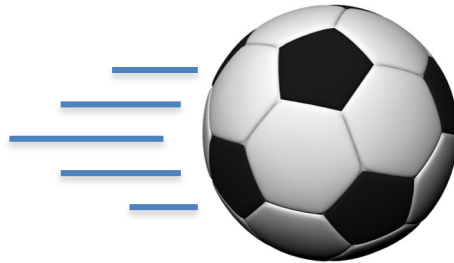
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"It is important to realize that in physics today, we have no knowledge what energy is."



Richard Feynman, in *The Feynman Lectures on Physics* (1964) Volume I, 4-1

Energy: a hint of history

The *vis viva* (living force), which **Gottfried Leibniz** defined as the product of the mass of an object and its velocity squared; he believed that total *vis viva* was conserved.



To account for slowing due to friction, Leibniz theorized that thermal energy consisted of **the random motion of the constituent parts of matter**, a view shared by **Isaac Newton**, although it would be more than a century until this was generally accepted.



Energy: a hint of history

The **conservation of energy** was proposed by [Gottfried Leibniz](#) over the period 1676–1689, the theory was controversial as it seemed to oppose the theory of [conservation of momentum](#) advocated by Sir [Isaac Newton](#) and [René Descartes](#). The two theories are now understood to be complementary.



In 1807, **Thomas Young** was possibly the first to use the term "energy" instead of vis viva, in its modern sense.



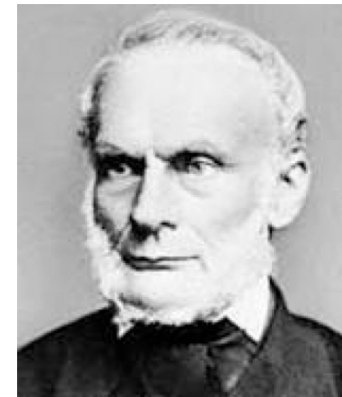
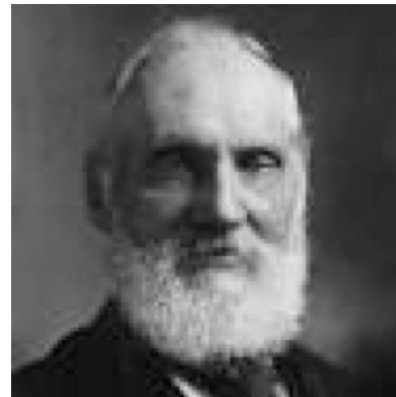
Energy: a hint of history

Gustave-Gaspard Coriolis described "kinetic energy" in 1829 in its modern sense.

In 1853, **William Rankine** coined the term "potential energy".

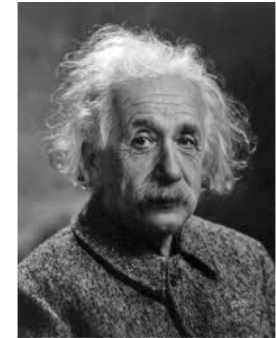
William Thomson (Lord **Kelvin**) amalgamated all of these laws into the laws of thermodynamics, which aided in the rapid development of explanations of chemical processes by

Rudolf Clausius, Josiah Willard Gibbs, and Walther Nernst.



Energy: a hint of history

It also led to a mathematical formulation of the concept of entropy by Clausius and to the introduction of laws of radiant energy by **Jožef Stefan**.



Albert Einstein proposed mass–energy equivalence in 1905 in a paper entitled "Does the inertia of a body depend upon its energy-content?".

Since 1918 it has been known that the law of conservation of energy is the direct mathematical consequence of the translational symmetry of the quantity conjugate to energy, namely time (**Emmy Noether**).

